Frankfurt, 27.11.2024

Are e-fuels a beneficial alternative to conventional fuels?



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definition

- fuels in generell
 - fuels: combustible substances → chemical energy is converted into mechanical energy through combustion in combustion engines.
- conventionel fuels (gasoline, diesel, kerosene)
 - a mixture of different hydrocarbons (HC) based on crude oil
- e-fuels (electrofuels)
 - synthetic fuels (also HC) that are produced from water and carbon dioxide (CO₂) using electrical energy
 - known as power-to-fuel (PtF or in general PtX)

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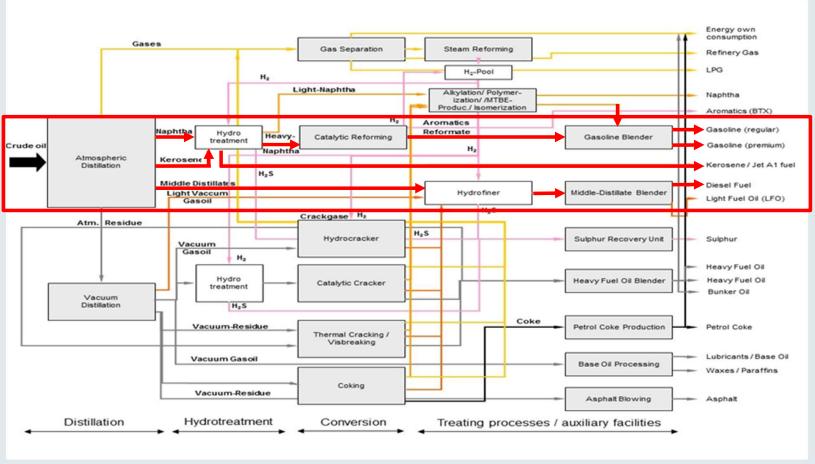


https://www.bmw.de





How conventional fuels are produced?





Why do we use hydrocarbons to generate energy?

- easy to handle
- no major hazard potential with careful handling
- high energy density

$$C_8H_{18} + 12,5 O_2 \rightleftharpoons 8 CO_2 + 9 H_2O_{(g)} + Energy$$



$$Energy = 5572 \, kJ/mol \equiv 13,55 \, kWh/kg \equiv 9,5 \, kWh/L$$

- hard coal
 - wood
 - natural gas

- 8,3 kWh/kg
- ≈4 kWh/kg
- ≈ 12,5 kWh/kg

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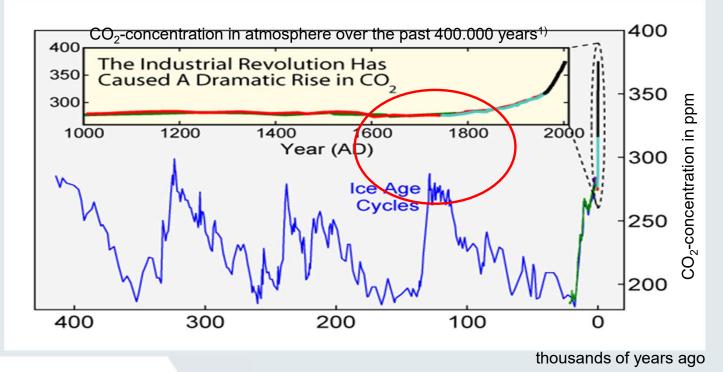
Die Zukunft unserer Energie

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What problems does the combustion of hydrocarbons cause?





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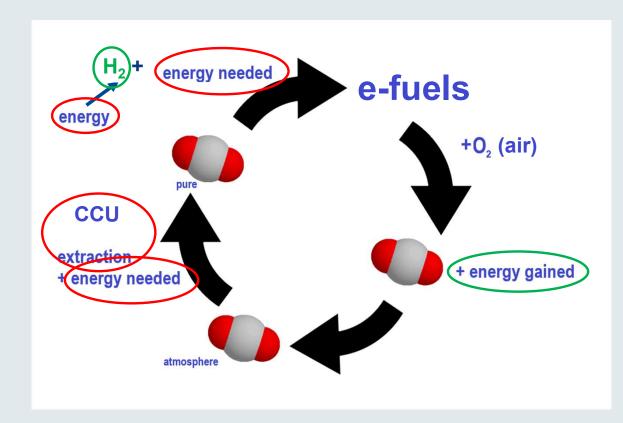
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Die Zukunft unserer Energie

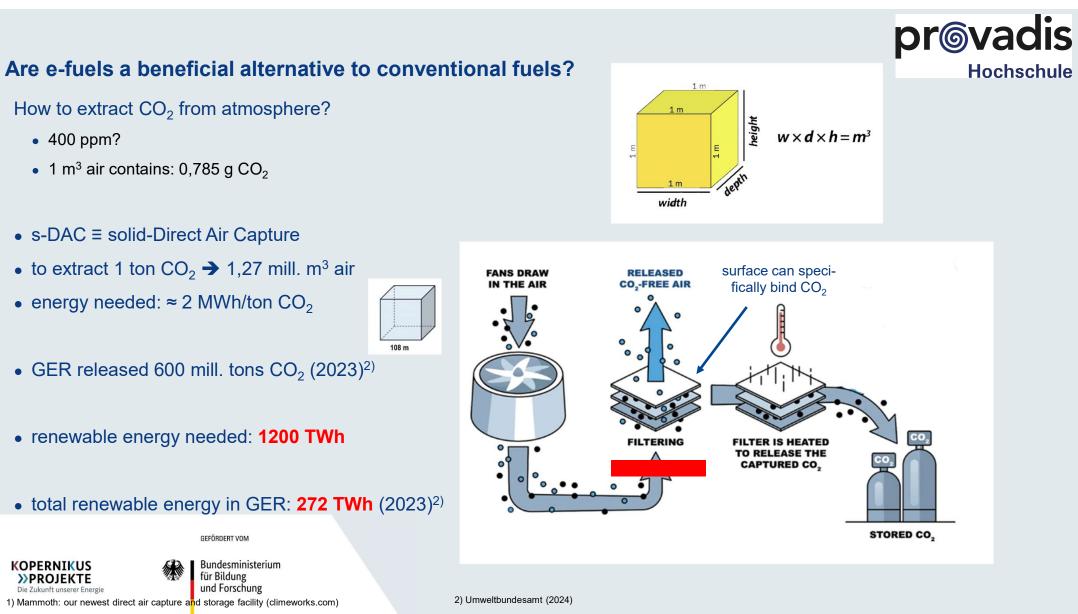
Bundesministerium für Bildung und Forschung 1) Hileman B: Ice Core Record Extended: Analyses of trapped air show current CO2 at highest level in 650,000 years. In: Chemical & Engineering News. Band 83, Nr. 48, November 2005, S. 7



How can we solve this problem?



- perfect carbon circle!
- CO_2 balance = 0
- but....
- 1. How to extract CO_2 ?
- 2. How to produce green hydrogen?
- 3. How much energy we need?
- 4. Are we in a position, to generate this energy in a climate-neutral way?
- 5. At what cost can e-fuels be produced?



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• 400 ppm?

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Die Zukunft unserer Energie

7



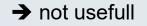
Are there other sources for CO₂ available?

- low concentration of CO₂
- hard coal-fired power station
- up to 15% CO₂ in exhaust gas
- why just 15%?
- purity of the exhaust gas?
- I-DAC ≡ liquid-Direct Air Capture
- energy consumption increases by 30%
- not all CO₂ could be captured
- coal-fired power generation will end in 2038 (Ger)



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Are e-fuels a beneficial alternative to conventional fuels?

Are there other sources for CO₂ available?

- low concentration of CO₂
- cement plant → cement production
 > responsible for 8% of global for CO₂ emissions¹⁾
- limestone is indispensable resource
 - ➤ responsible for 2/3 of emissions



https://www.rosenheim24.de/rosenheim/rosenheim-land/rohrdorf-ort50271/samstagoffenen-zementwerk-rohrdorf-rosenheim24-2362749.html

- polluted exhaust gas
- I-DAC ≡ liquid-Direct Air Capture
- energy needed: ≈ 2,4 MWh/ton CO₂²⁾



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https://www.spektrum.de/magazin/klimaneutrale-industrie-neuerfindung-von-zement/2150676
 geoengineeringmonitor.org (boell.de)
 Prof. Dr.-Ing. Ralf Ehret



How to produce green hydrogen?

• reverse watergas shift reaction (RWGS)

$$Energy + CO_2 + H_2 \xrightarrow{ca} CO + H_2O_{(g)}$$

- heterogeneously catalysed reaction
- several side reactions (Sabatier-reaction, methanisation, dry-, steamreforming, cocking,...)
- law of conservation of energy

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Die Zukunft unserer Energie

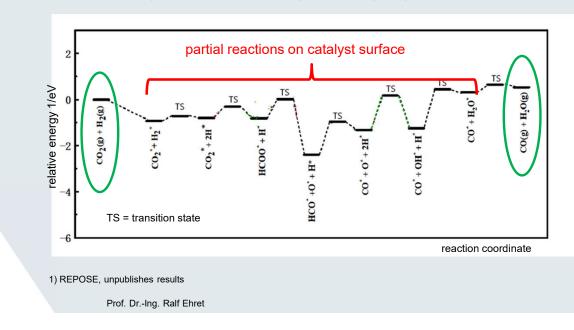
- included in Fischer-Tropsch-reaction
- energy needed: 0,85 MWh/ton CO¹⁾

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How to produce green hydrogen?

RWGS reaction → necessity of green hydrogen

$$Energy + CO_2 + H_2 \rightleftharpoons CO + H_2O_{(g)}$$



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• H₂O PEM-electrolysis Anode:
$$H_2O \rightleftharpoons 2H^+ + \frac{1}{2}O_{2(g)} + 2e^-$$

Proton Exchange Membrane Kathode: $2H^+ + 2e^- \rightleftharpoons H_{2(g)}$
dc-power source

$$Energy + H_2 0 \rightleftharpoons H_{2(g)} + \frac{1}{2}O_{2(g)}$$

• renewable energy needed: 50 MWh/ton H₂¹⁾

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KOPERNIKUS PROJEKTE Die Zukunft unserer Energie Bundesministerium für Bildung und Forschung 1) Frauenhofer IWS (2023)



https://www.siemens-energy.com/global/en/home/productsservices/product-offerings/hydrogen-solutions.html





SIEMENS Ingenuity for lif

How to produce green hydrogen?

- renewable energy needed: 50 MWh/ton H₂
- current H₂-demand in GER: 10^6 tons \equiv 50 TWh (almost industrial use)
- total renewable energy in GER: 272 TWh (2023)
- expected demand (2050¹): 15[•]10⁶ tons ≡ 750 TWh
- necessity of import → H₂ transportation





https://www.siemens-energy.com/global/en/home/productsservices/product-offerings/hydrogen-solutions.html





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1) https://www.bmbf.de/bmbf/de/forschung/energiewende-und-nachhaltiges-wirtschaften/nationalewasserstoffstrategie/nationale-wasserstoffstrategie_node.html#:~:



How to produce green hydrogen?

- How is the current situation in water electrolysis?
- currently 17 manufacturers worldwide offering 92 different systems (PEM, alkaline, high-temperature)



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- 2021: Chemiepark Rheinland (Wesseling) 10 MW PEM-electrolyser
 REFHYNE¹ → up to 1.300 tons H₂/y
- 8/2024: Oberhausen, 20 MW PEM-electrolyser TRAILBLAZER, Air Liquide
- 2024 Porsgrunn (Norway), 24 MW PEM-electrolyser (so far biggest in Europe)
- 2024 decision taken: 100 MW PEM-electrolyser REFHYNE II → up to 16.000 tons H₂/y (operational in 2027)
- 2023 Port of Rotterdam → start of construction of 5 plants, total capacity 1 GW
 → up to 180.000 tons H₂/y (operational in 2025-30)

1) Shell Deutschland Oil GmbH, SINTEF, Element Energy and Sphera

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How to produce green hydrogen?

- projekt: OffsH2ore¹⁾
- feasibility study
 - off-shore wind farm
 - 500-MW platform for H₂O-PEM-electrolyses
 - capacity: 50.000 tons/y H₂
 - modular concept → easy scalability
 - energy: reversed osmosis → waste heat from electrolyses
 - clean and dry H₂ → compressed to 500 bar
- transportation:
 - via pipeline
 - via ship (max load 400 tons)

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aufgrund eines Beschlusses des Deutschen Bundestages



Schematic illustration of an offshore concept for hydrogen production and pressurised gas transport (ship or pipeline)

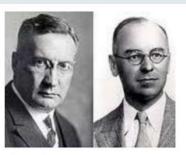
	ecology (LCA)	economy (TEA)			
pipeline	+	-			
ship	-	+			

1) PINE AG, Fraunhofer-Institut für Solare Energiesysteme ISE, SILICA Verfahrenstechnik GmbH, KONGSTEIN GmbH



How to produce e-fuels?

- syn gas: CO (from CO₂ via RWGS) + H₂
- Fischer-Tropsch-reaction:
- developed 1920 → commercialized in 1930s
- historical: coal liquefaction (2 steps)
- per kg HC \rightarrow 1,25 kg H₂O \rightarrow loss of H₂
- side products (CA, alcoholes, aldehydes, ketons, ...)



Franz Fischer Hans Tropsch

www.wikipedia.de



SASOL - Slurry phase distillate reactor¹⁾

1) https://websites.umich.edu/~elements/fogler&gurmen/html/01chap/html/reactors/sasol.htm

 $n CO + 2n H_2 \xrightarrow{cat.} HC + m H_2O_{(g)} + Energy$

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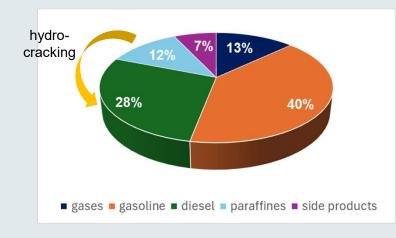
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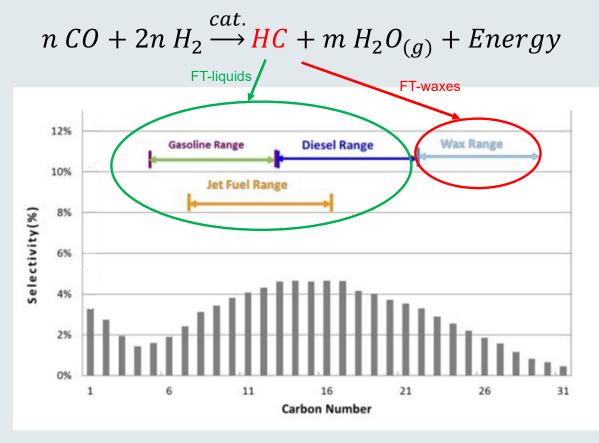
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How to produce e-fuels?

- Fischer-Tropsch-(FT)-reaction
- what means "HC"?
- product distribution = f (T, p, cat., reactor system, t)
- distribution for L(ow)T(emp.)-FT:

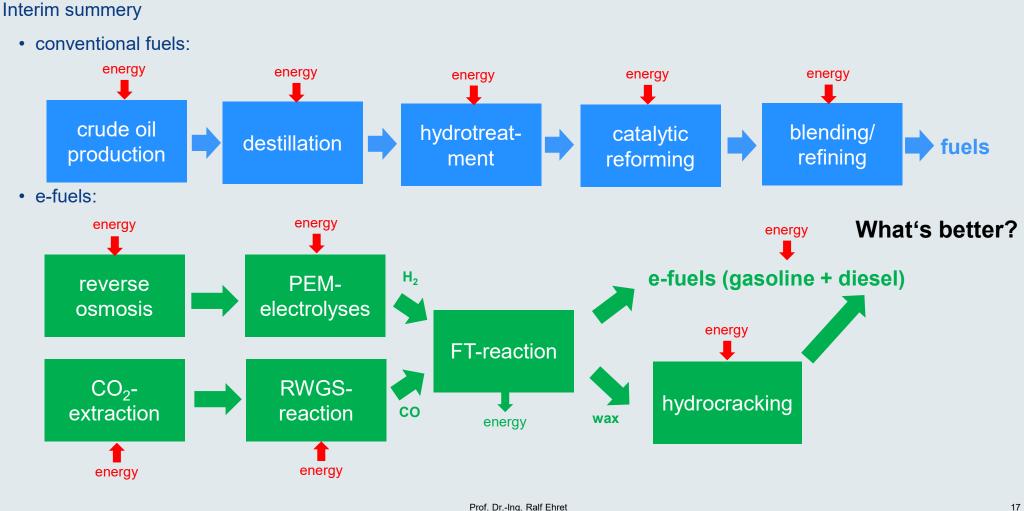




product distribution of LTFT (200 - 250 °C)¹⁾

1) Rui Xu, Departement of Chemical Engineering, Auburn Univ, 2013







Energy needed?

• to produce 1 L fuel (gradle to tank)

 conventional fuel 	1,6 - 1,8 kWh ¹⁾	≈2 kWh
• e-fuel	18 - 25 kWh ¹⁾	≈ 22 kWh

• fuel consumption in GER 2023²)

fuel	quantity [tons]	density [kg/L]	volume [m ³]
diesel	36*10 ⁶	0,84	42,9*10 ⁶
gasoline	18*10 ⁶	0,74	24,3*10 ⁶
kerosene	9*10 ⁶	0,80	11,3*10 ⁶
total	63*10 ⁶		

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1) Frauenhofer IWS 2020, Deutsches Zentrum für Luft- und Raumfahrt, 2021



, Arbeitsgemeinschaft Energiebilanzen e.V.

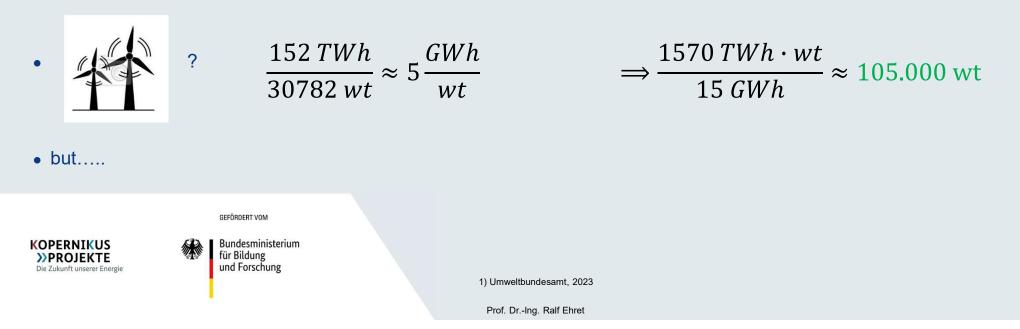


Additional energy needed?

 $Energy = 78,5 \cdot 10^{9}L \cdot 20 \ kWh/L = 1.570 \cdot 10^{9} \ kWh \equiv 1.570 \ TWh$

• renewable energy available in GER¹⁾

 $Energy (2023) = 513 TWh \Rightarrow 53\% renewable \Rightarrow 272 TWh$

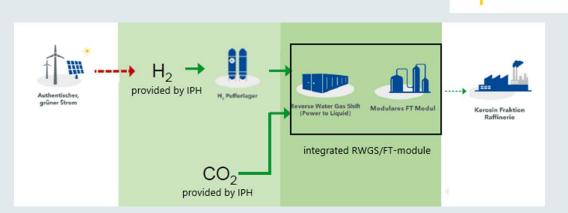




Use of excess energy!

Energy lost $(2023)^{1} = 8.6 TWh \equiv 430.000 m^3 fuel$

- large scale FT-plants in Germany? → small scale
- REPOSE²⁾ ≡ Renewable Power Supply for e-fuels
- capacity: 2.500 tons/y fuel





INERATEC plant at IPH (operational Dec. 24)

1) PV- and wind energy, Umweltbundesamt, 2023, Statistisches Bundesamt 2023

2) CENA, Provadis Hochschule, Ineratec, Frauenhofer Institut

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für Verkehr und digitale Infrastruktur

Transportation

- fuels (diesel, gasoline, kerosene) and CO_{2(sc/l)} → pipelines oder ships → established technology
- H₂?

energy source	calorific value [kWh/kg]	density [kg/m³]	volume [L]	in relation to cv
H ₂	33	0,09	11.200	3735
gasoline	11	750	1,3	1,3

• liquification

energy source	calorific value [kWh/kg]	density [kg/m³]	volume [L]	in relation to cv
H ₂ (liquid)	33	71	14,1	4,7
gasoline	11	750	1,3	1,3





Transportation

- boiling point H₂?
- storage, loading, loss of H₂ during transportation via ship.....?
- compression?

https://www.chemie-master.de/FrameHandler.php?loc=https://www.chemie-master.de/pse/pse.php?modul=O

increases with temperature and pressure

energy source [600 bar]	calorific value density [kWh/kg] [kg/m³]		volume [L]	in relation to cv
H ₂	33	48	18,7	6,2
gasoline	11	750	1,3	1,3

hydrogen induced embrittlement → cracking





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- → specific steel alloys or composite reinforced pipelines
- transportation of H₂ is a major challenge!



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Other possibilities für H₂-transportation?

- as a chemical compound → e.g. NH₃
- March 2023: Signing of feasibility and implementation agreement between Namibia and Hyphen Hydrogen Energy¹⁾
- to proof within 2 years the possibility to install that:
- total invest: 10 billion \$ →
- desaltination plant
- H₂O-electrolyses → H₂
- Linde air separation plant \rightarrow N₂
- Haber-Bosch-ammonia plant → NH₃
- new harbour at Lüderitz bay + infrastructur
- capacity/year: 2 million tons ammonia, 350.000 tons H₂, 7.000 MW PV-electricity (total Namibia: 600 MW)
- transportation on liquid $NH_3 \rightarrow$ boiling point: -33 °C (H_2 : -252 °C)
- conversion into $H_2 \rightarrow$ via thermal catalytic cracking



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1) joint venture between Nicholas Holdings Limited and ENERTRAG AG





model of the planned plant in Namib Desert¹⁾



https://www.google.de/maps



The life cycle assessment view

• kg CO₂-equivalents for 1 L fuel

	convent	tional fuels	e-fuels		
	gasoline diesel		gasoline	diesel	
well to tank (WTT) ¹⁾	0,5 - 1	0,6 - 1,2	0 - 2,4	0 - 2,7	
tank to wheel (TTW) ²⁾	2,4	2,7	0	0	
total (WTW)	2,9 - 3,4	3,3 - 4,5	0 - 2,4	0 - 2,7	

1) indirect emissions \rightarrow extraction, production + rafination, transport

2) direct emissions \rightarrow combustion

• 100% renewable energy is a necessity



			only costs for e-fuel production		
Tł	ne techno-economic view		N		
•	tax & OPEX values for 1	L gasoline			
		conventional fuel	e-fuel		
		1,60 €		3,18€	
	mineral oil tax1)	65 ct	mineral oil tax ¹	65 ct	
	CO ₂ tax	10 ct	CO ₂ tax	-	
		85 ct		253 ct	
	crude oil		electricity + CO_2	20 ct	
	refinery		hydrogen	180 ct	
	transportation		transportation	11 ct	
	margin		margin	15 ct	
	VAT		VAT	27 ct	

1) fix amount, no percentage, diesel mineral oil tax: 47 ct

2) Promotionsarbeit S. Schemme, FZ Jülich 2020



Current situation?

"Ørsted scraps flagship European green fuels project"

- 201 mil. € investment
- 55.000 tons green Methanol (MeOH)
- CO₂ → combined heat and power plant + green H₂
- cancelled in 8/2024
- MeOH price to high

"Fulcrum BioEnergy abandons trash-to-fuel plant in Nevada. The waste gasification start-up abruptly laid off most staff in mid-May" (2024)

- capacity 42 million L of e-fuel per year
- employed about 120 people.
- permits no extended and technical problems



Kurz vor Fertigstellung stoppt der dänische Konzern den Bau einer Fabrik für synthetisches Methanol. Der Grund: Es rechnet sich nicht mehr.



https://cen.acs.org/energy/Fulcrum-BioEnergy-abandons-trashfuelplant/102/web/2024/06

Shell Nederland Raffinaderij B.V is to temporarily pause on-site construction work at its 820,000 tonnes a year biofuels facility at the Shell Energy and Chemicals Park Rotterdam in the Netherlands to address project delivery and **ensure future competitiveness given current market conditions** (July 2024).

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Are e-fuels a beneficial alternative to conventional fuels?

Current situation in Ger?



	Projekt	Hauptprodukt	Kapazität	Standort	Produktionstechnologie	Angekündigte Inbetriebnahme	CO2-Quelle	Geplantes Einsatzfeld
1	Solarbelt	Kerosin	360 t/a	Werlte	Fischer-Tropsch	Seit 2021	Punktquelle Biomethan	Luftfahrt
2	NextGate	eFuels und Wachse	200 t/a	Hamburg	Fischer-Tropsch	Seit 2022	Biogenes CO2 per Tankwagen	Keine Angabe
3	Reallabor Westküste 100/ KeroSyn100	Kerosin	600 t/a	Demonstrationsanlage Raffinerie Heide	Methanol-to-Jet	2023/2024	Punktquelle Zementwerk	Luftfahrt
4	Ineratec	eFuels	3.500 t/a	Industriepark Hoechst	Fischer-Tropsch	2024	Biogas	Keine Angabe
5	Shell Rheinland Raffinerie	Kerosin	100.000 t/a1	Köln	Nicht bekannt	2025	Altholz	Luftfahrt
6	HyKero	Kerosin	42.000 t/a	Böhlen-Lippendorf	Fischer-Tropsch	2026	Punktquelle Biomethan	Luftfahrt u.a.
7	Jangada	Kerosin	34.000 t/a	ehemaliger Flugplatz in Drewitz (GRAL)	Fischer-Tropsch	2027	Biogene Punktquelle	Luftfahrt
8	Reallabor Westküste 100/ KeroSyn100	Kerosin	20.000 t/a	Demonstrationsanlage Raffinerie Heide	Methanol-to-Jet	2027	Punktquelle Zementwerk	Luftfahrt
9	reFuels – Kraftstoffe neu denken	Raffination Syn- Crude	Bis zu 50.000 t/a	MiRO Karlsruhe	Fischer-Tropsch	2027	Nicht bekannt	Nicht bekannt
10	OMV Burghausen	Kerosin	50.000 t/a	Voraussichtlich im Raum Burghausen (Bayern)	Nicht bekannt	Ende der 2020er- Jahre	Nicht bekannt	Luftfahrt
11	Concrete Chemicals	Kerosin u.a.	30.000 t/a SAF	Rüdersdorf	Fischer-Tropsch	2028	Punktquelle Zementwerk	Luftfahrt u.a.
12	DAWN	Kerosin	3.500 t/a	Jülich	Fischer-Tropsch	Nicht bekannt	Sun to liquid	Luftfahrt
13	E-Kerosin-aus-der-Luft	Kerosin	274 t/a	Rostock- Laage	Fischer-Tropsch	Nicht bekannt	Nicht bekannt	Luftfahrt u.a.
14	Technologie-Plattform (TPP)	SynCrude, Kero- sin, Methanol u.a.	10.000 t/a	Leuna	Diverse Forschungsmodule	Baubeginn: 2024	Nicht bekannt	Keine Angabe
15	Green Fuels Lausitz	Methanol und Kerosin	200.000 t/a	Industriepark Schwarze Pumpe (Lausitz)	Methanol-Synthese/Fischer-Tropsch	Nicht bekannt	Abfallverwertung	Keine Angabe
16	Green MeOH	Methanol	200.000 t/a	Chemiepark in Stade	Methanol-Synthese	Nicht bekannt	Punktquelle Gaskraftwerk	Seefahrt
17	PtX 1.0	Kerosin	274 t/a	Böhlen- Lippendorf	Nicht bekannt	Nicht bekannt	Nicht bekannt	Luftfahrt
18	PtX Lab Lausitz	Kerosin	10.000 t/a	Lausitz	Fischer-Tropsch	In Planung	Biogene Punktquelle/DAC	Luftfahrt
19	SAF@STR	Kerosin	120.000 t/a	Stuttgart	Nicht bekannt	Nicht bekannt	Punktquelle Zementwerk	Luftfahrt

Dozent:in



summary

- extreme amounts of renewable electrical energy required
 - so far not available → heavy investment necessary
- an overnight changeover is not feasible
 - stepwise approach → blending of e-fuels via a quota system and/or taxes
- challenging H₂ transportation
 - no sufficient solution so far → more research needed
- economy vs. ecology
 - politics can set the framework conditions through taxes
- fully integrative systems (DAC + elektrolyses → e-fuels) necessary
 - outside GER → continued dependencies
- economy of scale

E-fuels as an interim solution?

Why e-fuels at all?



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Your opinion is important - We look forward to your participation!

Ihre Meinung ist wichtig - Wir freuen uns über Ihre Teilnahme!



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